

Saturday Morning of Theoretical Physics, 17 November 2018



ENTROPY:

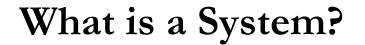
Gaining Knowledge by Admitting Ignorance

Alexander Schekochihin

Rudolf Peierls Centre for Theoretical Physics University of Oxford

with thanks to James Binney, who convinced me that it's better to gamble than to count







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 $\alpha = 1, 2, 3, \ldots, \Omega \gg 1$



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Mean energy:
$$U = \langle E_{\alpha} \rangle = \sum_{\alpha} p_{\alpha} E_{\alpha}$$

Pressure: $P = -\left\langle \frac{\partial E_{\alpha}}{\partial V} \right\rangle = -\sum_{\alpha} p_{\alpha} \frac{\partial E_{\alpha}}{\partial V}$
(volume)



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More generally, Force $= -\sum_{\alpha} p_{\alpha} \frac{\partial E_{\alpha}}{\partial \text{Displacement}}$



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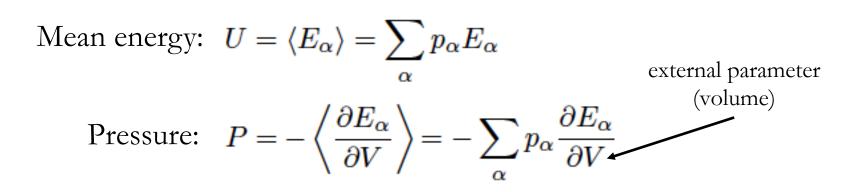
If we know p_{α} , we can calculate averages \leftrightarrow predict measurement outcomes

Mean energy:
$$U = \langle E_{\alpha} \rangle = \sum_{\alpha} p_{\alpha} E_{\alpha}$$

E.g., tension: $f = \sum_{\alpha} p_{\alpha} \frac{\partial E_{\alpha}}{\partial L}$, magnetisation: $M = -\sum_{\alpha} p_{\alpha} \frac{\partial E_{\alpha}}{\partial B}$

What Do We Want to Know?



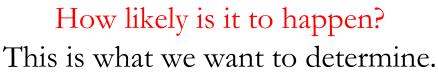


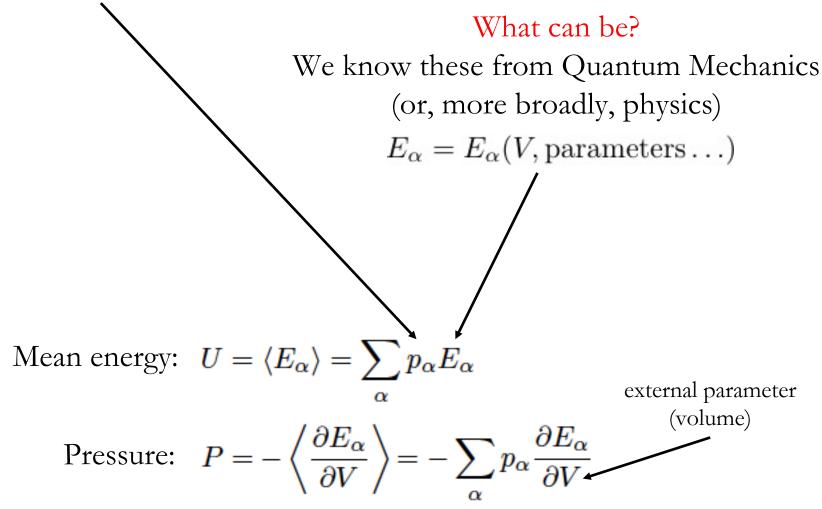
What Do We Want to Know?



What can be? We know these from Quantum Mechanics (or, more broadly, physics) $E_{\alpha} = E_{\alpha}(V, \text{parameters}...)$ Mean energy: $U = \langle E_{\alpha} \rangle = \sum p_{\alpha} E_{\alpha}$ external parameter (volume) Pressure: $P = -\left\langle \frac{\partial E_{\alpha}}{\partial V} \right\rangle = -\sum p_{\alpha} \frac{\partial E_{\alpha}}{\partial V}$









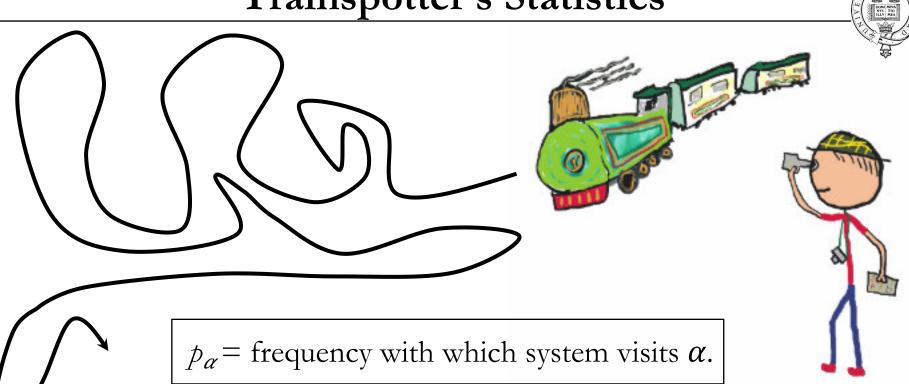
How likely is it to happen? This is what we want to determine.

We do not need to know the p_{α} 's in any "true" sense. We just need to know such p_{α} 's that the few macroscopic quantities that we are ever likely to measure come out right.

Mean energy:
$$U = \langle E_{\alpha} \rangle = \sum_{\alpha} p_{\alpha} E_{\alpha}$$

Pressure: $P = -\left\langle \frac{\partial E_{\alpha}}{\partial V} \right\rangle = -\sum_{\alpha} p_{\alpha} \frac{\partial E_{\alpha}}{\partial V}$ (volume)

Trainspotter's Statistics

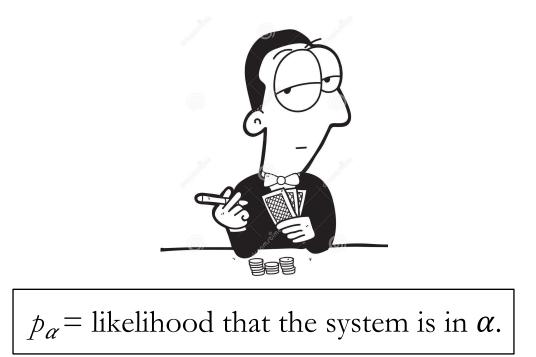


Technically speaking, to know this, we need to understand (or assume something about) the system's dynamics.

Also, we must believe that the time necessary for the system to visit all of its phase space is a reasonable time for us to wait.

This is not, however, how forecasting works...





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$$p_{\alpha} = \frac{1}{\varOmega}$$

(all states are equally probable)





Ludwig Boltzmann (1844-1906)

 p_{α} = likelihood that the system is in α .

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> $p_{\alpha} = rac{1}{\Omega}$ isolated system in equilibrium (Boltzmann's equal a priori probabilities postulate)





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A fair and adequate assignment of likelihoods must be based on some information about the system, however incomplete. What if we know one thing, e.g., mean energy:

$$U = \langle E_{\alpha} \rangle = \sum_{\alpha} p_{\alpha} E_{\alpha}$$

This is one constraint on $\overline{\Omega}$ >>>>1 likelihoods. How do we work out p_{α} 's using this and *only this* information?

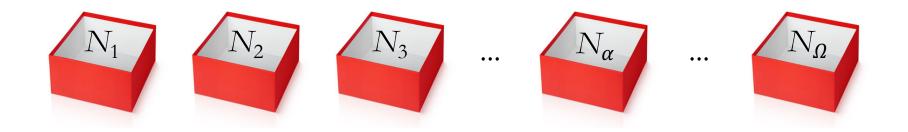


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1) Randomly assign $N >> \Omega$ "units of likelihood" to each of Ω states: N_{α} to state α . Then

$$p_{\alpha} = \frac{\mathcal{N}_{\alpha}}{\mathcal{N}}$$



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The most likely outcome of this game is the one that can be obtained in the largest number of ways:

this maximum is very sharp at large N

$$W = \frac{\mathcal{N}!}{\mathcal{N}_1! \cdots \mathcal{N}_{\Omega}!} \to \max \text{ (subject to constraints)}$$

(thus, we determine p_{α} 's by maximising W in this way)



Using Stirling's formula $\ln \mathcal{N}! = \mathcal{N} \ln \mathcal{N} - \mathcal{N} + O(\ln \mathcal{N})$, get:

$$\ln W = \underbrace{\mathcal{N} \ln \mathcal{N} - \mathcal{N}}_{\sum_{\alpha} (\mathcal{N}_{\alpha} \ln \mathcal{N} + \mathcal{N}_{\alpha})} + O(\ln \mathcal{N}) - \sum_{\alpha} [\mathcal{N}_{\alpha} \ln \mathcal{N}_{\alpha} - \mathcal{N}_{\alpha} + O(\ln \mathcal{N}_{\alpha})]$$
$$= -\sum_{\alpha} \mathcal{N}_{\alpha} \ln \frac{\mathcal{N}_{\alpha}}{\mathcal{N}} + O(\ln \mathcal{N})$$
$$= -\mathcal{N} \left[\sum_{\alpha} p_{\alpha} \ln p_{\alpha} + O\left(\frac{\ln \mathcal{N}}{\mathcal{N}}\right) \right].$$

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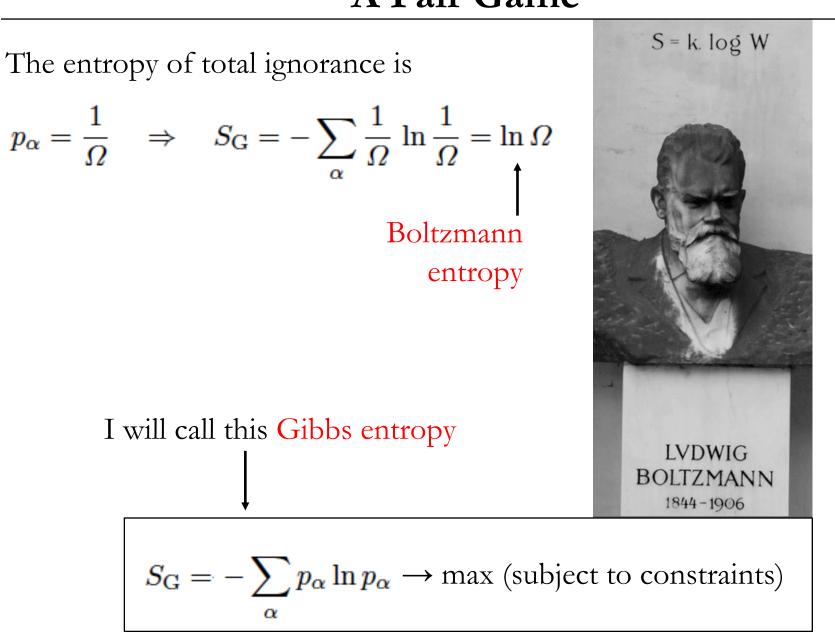
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I will call this Gibbs entropy
$$\int J_{\text{Osiah Willard Gibbs}} J_{\text{Osiah Willard Gibbs}} (1839-1903)$$

$$S_{\text{G}} = -\sum_{\alpha} p_{\alpha} \ln p_{\alpha} \rightarrow \max \text{ (subject to constraints)}$$

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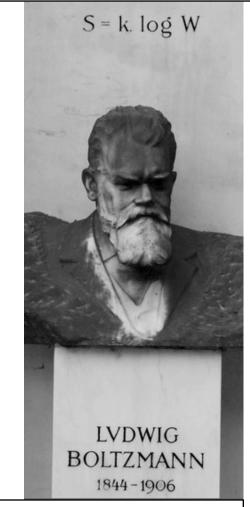
The entropy of total ignorance is

$$p_{\alpha} = \frac{1}{\Omega} \quad \Rightarrow \quad S_{\rm G} = -\sum_{\alpha} \frac{1}{\Omega} \ln \frac{1}{\Omega} = \ln \Omega$$

It corresponds to the number of all possible distributions of N units of probability between Ω states: $W_{\text{max}} = \Omega^N$, so

$$S_{\rm G,max} = \frac{1}{\mathcal{N}} \ln W_{\rm max} = \ln \Omega$$

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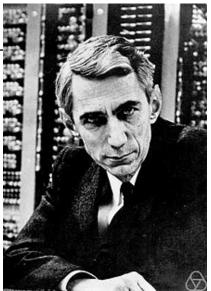
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Shannon declared that a good measure of uncertainty of set of probabilities p_{α} , called $H(p_1, \ldots, p_{\Omega})$, must be 1) a continuous function of p_{α} 's

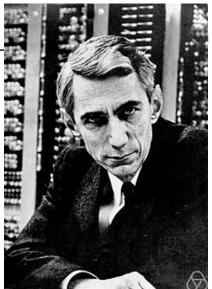


Claude Shannon (1916-2001)

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Shannon declared that a good measure of uncertainty of set of probabilities p_{a} , called $H(p_1, \ldots, p_{\Omega})$, must be

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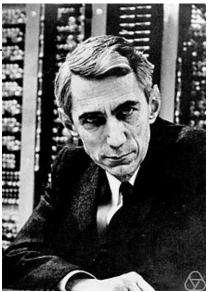
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$$H(p_1,\ldots,p_\Omega) < H\left(\frac{1}{\Omega},\ldots,\frac{1}{\Omega}\right) \equiv H_\Omega$$



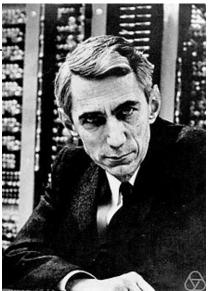
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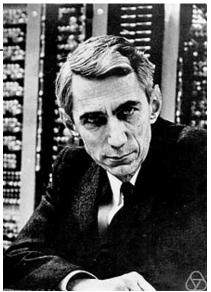
4) more equiprobable states \rightarrow more uncertainty: if $\Omega' > \Omega$, $H_{\Omega'} > H_{\Omega}$

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5) additive and independent of counting scheme: if we split all states into subgroups with probabilities w_i, and calculate H_i for each subgroup (with conditional probabilities p_α/w_i), then

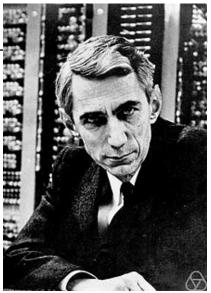
$$\begin{array}{ll} H(p_1, \dots, p_{\Omega}) = H(w_1, w_2, \dots) + \sum_i w_i H_i \\ \text{uncertainty} & \text{uncertainty} \\ \text{uncertainty} & \text{in choosing subgroup} & \text{within subgroup} \end{array}$$

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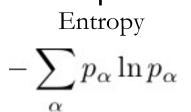
He then proved that the only function with these properties is

$$H(p_1,\ldots,p_\Omega) = -k\sum_{\alpha} p_{\alpha} \ln p_{\alpha} = kS_G$$

How to Maximise Entropy (with strings attached)

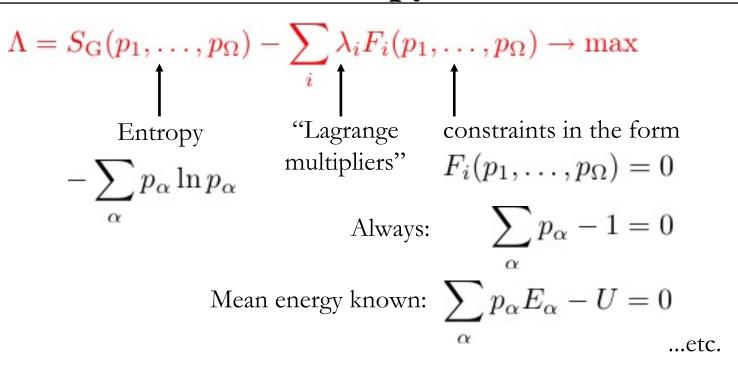






"Lagrange constraints in the form multipliers" $F_i(p_1, \ldots, p_\Omega) = 0$ How to Maximise Entropy (with strings attached)









$$\Lambda = -\sum_{\alpha} p_{\alpha} \ln p_{\alpha} - \lambda \left(\sum_{\alpha} p_{\alpha} - 1 \right) \to \max$$

Always:
$$\sum_{\alpha} p_{\alpha} - 1 = 0$$

Mean energy known:
$$\sum_{\alpha} p_{\alpha} E_{\alpha} - U = 0$$
...etc

E.g., in the case of **total ignorance**, the only constraint is

$$\frac{\partial \Lambda}{\partial p_{\alpha}} = -\ln p_{\alpha} - 1 - \lambda = 0 \quad \Rightarrow \quad p_{\alpha} = e^{-(1+\lambda)} \quad \text{all equal}$$

Total Ignorance (Isolated System)



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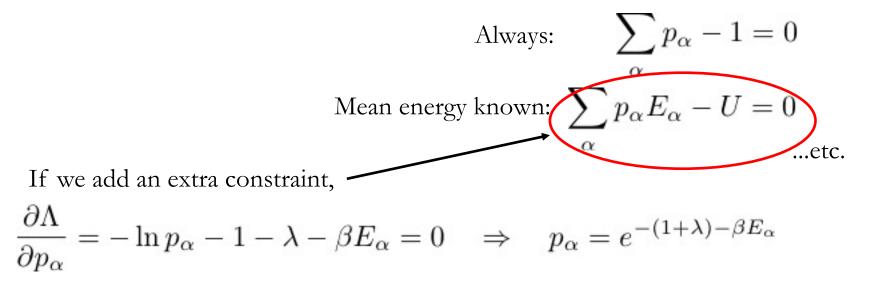
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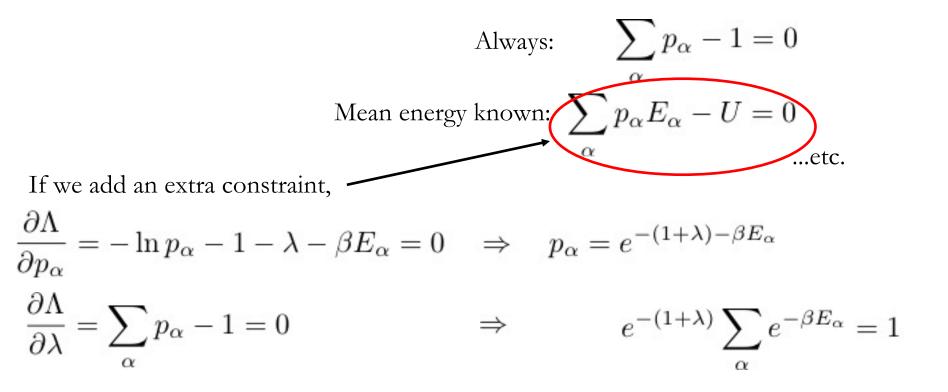


$$\Lambda = -\sum_{\alpha} p_{\alpha} \ln p_{\alpha} - \lambda \left(\sum_{\alpha} p_{\alpha} - 1 \right) - \beta \left(\sum_{\alpha} p_{\alpha} E_{\alpha} - U \right) \to \max$$



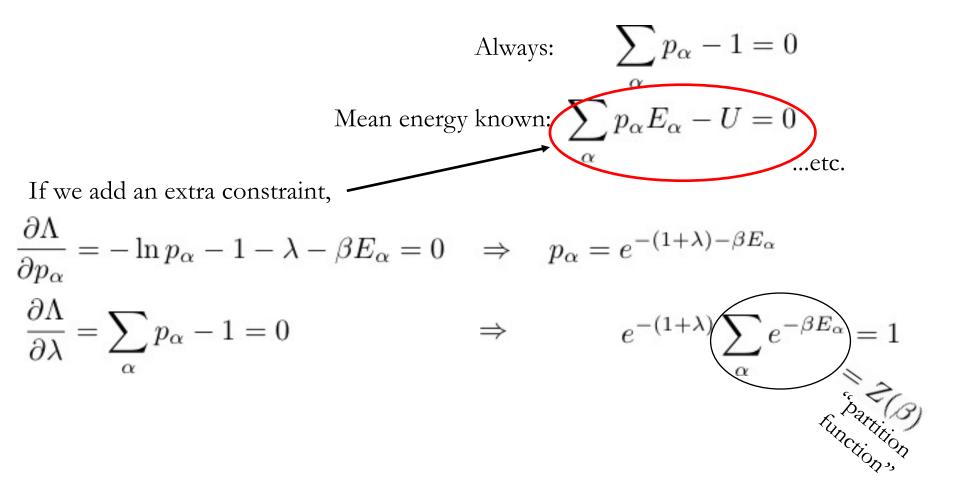


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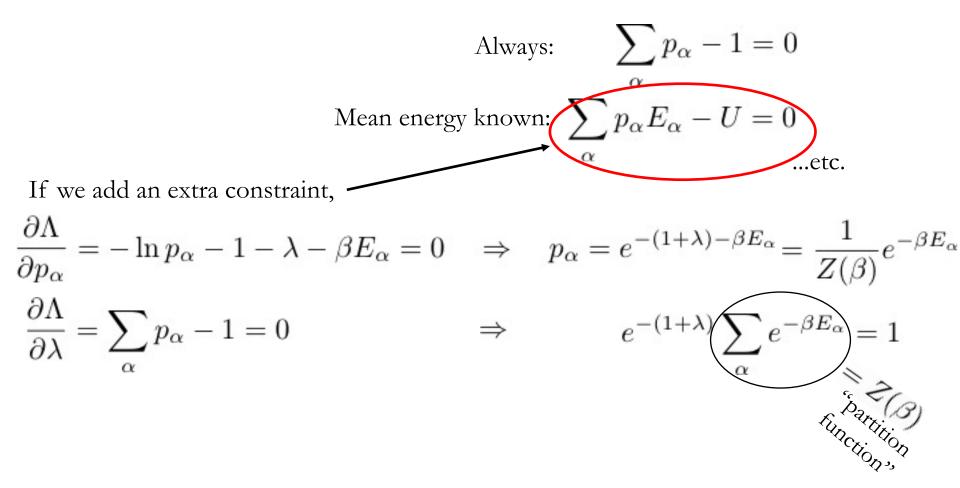


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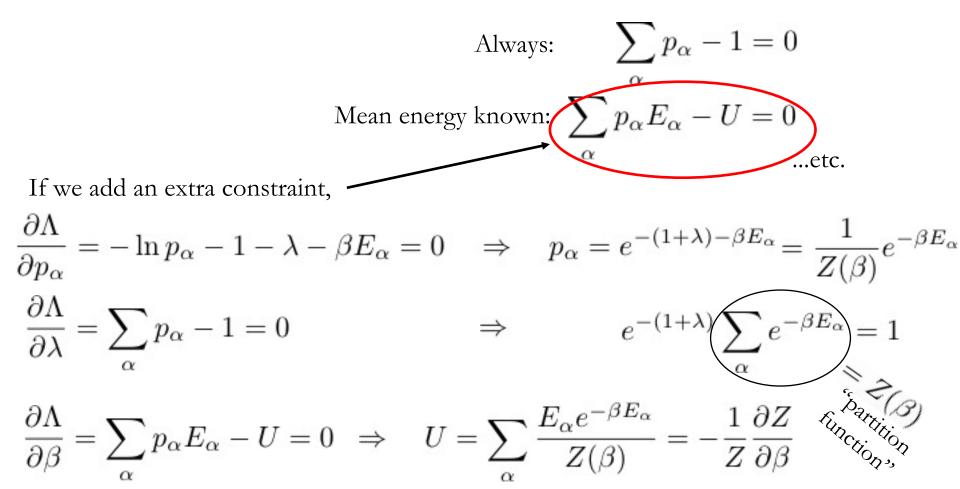


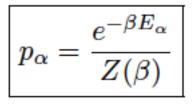
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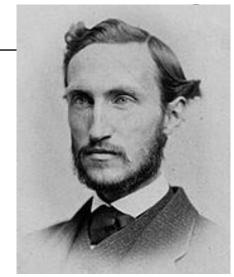
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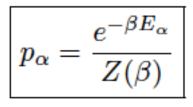


likelihoods based of knowledge of \boldsymbol{U}

$$Z(\beta) = \sum_{\alpha} e^{-\beta E_{\alpha}} \text{ normalisation constant}$$
$$U = -\frac{1}{Z} \frac{\partial Z}{\partial \beta} \text{ implicit equation for mysterious } \beta$$

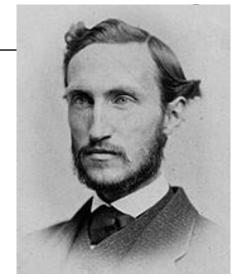


Josiah Willard Gibbs (1839-1903)



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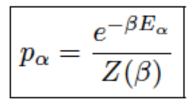
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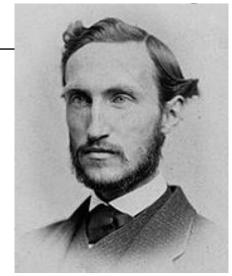
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$$dS_{\mathbf{G}} = \beta dU + U d\beta + \frac{dZ}{Z} \quad \text{``d'' here means we are varying some}$$
parameter of the equilibrium, say V



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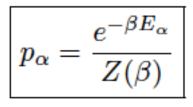


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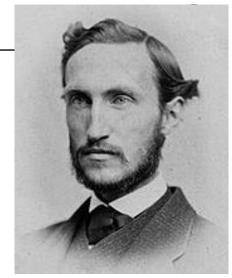
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$$= \beta dU + U d\beta + \sum_{\alpha} \frac{e^{-\beta E_{\alpha}}}{Z} (-\beta dE_{\alpha} - E_{\alpha} d\beta)$$



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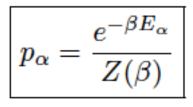


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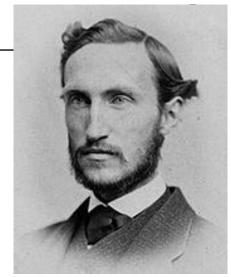
$$dS_{\rm G} = \beta dU + U d\beta + \frac{dZ}{Z} \qquad \qquad dE_{\alpha} = \frac{\partial E_{\alpha}}{\partial V} dV$$

$$= \beta dU + U d\beta + \sum_{\alpha} \underbrace{\frac{e^{-\beta E_{\alpha}}}{Z}}_{= p_{\alpha}} (-\beta dE_{\alpha} - E_{\alpha} d\beta)$$



likelihoods based of knowledge of \boldsymbol{U}

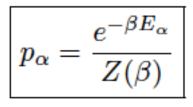
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Josiah Willard Gibbs (1839-1903)

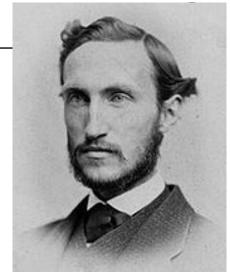
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likelihoods based of knowledge of U

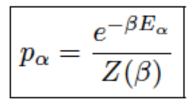
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likelihoods based of knowledge of \boldsymbol{U}

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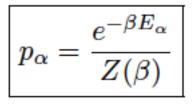
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 $\mathrm{d}S_{\mathrm{G}} = \beta \,\mathrm{d}Q_{\mathrm{rev}}$

Thus, change of entropy is proportional to reversible heat, and the const of proportionality is β .

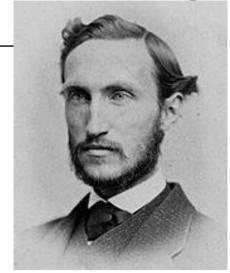


Entropy Is "Real"!



likelihoods based of knowledge of $\,U\,$

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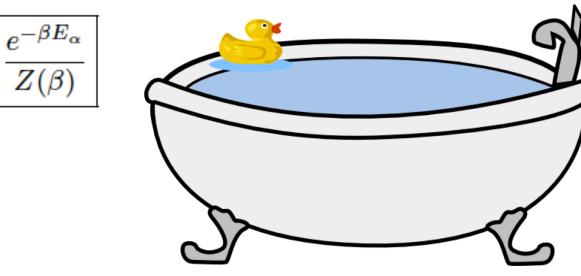
$$\mathrm{d}S_{\mathbf{G}} = \beta \,\mathrm{d}Q_{\mathrm{rev}}$$

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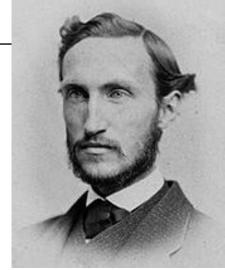
$$\mathrm{d}S = \frac{\mathrm{d}Q_{\mathrm{rev}}}{T}$$

 $\beta = \frac{1}{k_{\rm B}T}$ and $S = k_{\rm B}S_{\rm G}$ temperature Gibbs entropy is thermodynamic entropy!

System in a Bath



 $p_{\alpha} =$



Josiah Willard Gibbs (1839-1903)

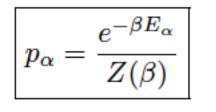
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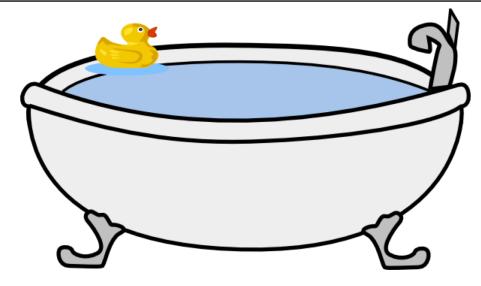
Physically, a system at fixed temperature is a system embedded in a "thermal bath",

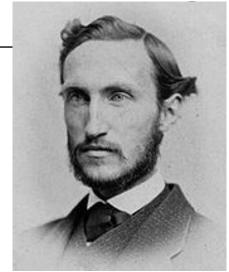
i.e., a much larger environment with which it can exchange energy

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 and $S = k_{\rm B}S_{\rm G}$
temperature Gibbs entropy is
thermodynamic entropy!

System in a Bath







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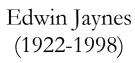
more fun and games with thermodynamics in J. Chalker's lecture



The thermodynamic entropy of the world (closed system) increases with time.



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Probability Theory The Logic of Science

CAMBRIDGE

E. T. JAYNES

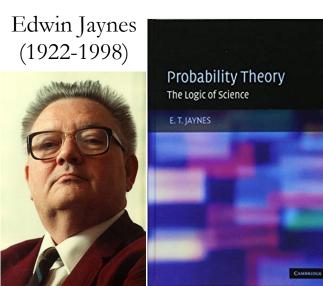




The thermodynamic entropy of the world (closed system) increases with time.

Time t: Measure something about some part of the world, maximise S_G subject to that info, get a set of p_{α} 's and

 $S(t) = k_{\rm B} S_{\rm G,max}(t)$





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$$S(t') = \underbrace{k_{\rm B}S_{{\rm G},{\rm max}}(t')}_{\text{the new }S_{{\rm G}},} \quad \geqslant \quad \underbrace{k_{\rm B}S_{{\rm G}}(t')}_{\text{the "true"}} = k_{\rm B}S_{{\rm G}}(t) = S(t)$$

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The thermodynamic entropy of the world (closed system) increases with time.

The increase of S reflects our insistence to forget the detailed knowledge that we possess as a result of evolving in time any earlier state (even if based on earlier statistical inference) and re-apply at every later time the rules of statistical inference based on knowledge obtained in new measurements.



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S = entropy of the object of measurement + entropy of observer & her kit.
 A very precise measurement will imply massive increase in the latter.
 (More generally, the system gets entangled with its environment and when we measure it, our ignorance of the latter pollutes our knowledge of the former.)



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go seriously quantum with Sid Parameswaran